

ω6-OXYGENATION OF 6,9,12-OCTADECATRIENOIC ACID  
IN HUMAN PLATELETS

Mats Hamberg

Department of Chemistry  
Karolinska Institutet, S-104 01 Stockholm 60, Sweden

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**Summary:** [1-<sup>14</sup>C]6,9,12-Octadecatrienoic acid was incubated with suspensions of human platelets. Three monohydroxy acids were isolated, i.e. 10L<sub>5</sub>-hydroxy-6,8-pentadecadienoic acid, 10L<sub>5</sub>-hydroxy-6,8,12-octadecatrienoic acid, and 13L<sub>5</sub>-hydroxy-6,9,11-octadecatrienoic acid. Aspirin (0.5 mM) and indomethacin (10 μM) completely inhibited formation of the first mentioned compound whereas 5,8,11,14-eicosatetraenoic acid (34 μM) inhibited formation of all three compounds. Isolation of 13L<sub>5</sub>-hydroxy-6,9,11-octadecatrienoic acid demonstrates that human platelets possess a lipoxygenase activity catalyzing ω6-oxygenation of suitable poly-unsaturated fatty acids.

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Arachidonic acid incubated with human platelets is transformed into 12L<sub>5</sub>-hydroxy-5,8,10-heptadecatrienoic acid and thromboxane A<sub>2</sub> (1; cyclooxygenase pathway) and into 12L<sub>5</sub>-hydroxy-5,8,10,14-eicosatetraenoic acid (1,2; lipoxygenase pathway). The substrate specificity of platelet lipoxygenase was studied by Nugteren (2). He found that a number of C<sub>20</sub> polyenoic acids as well as 6,9,12-octadecatrienoic acid and 4,7,10,13,16,19-docosahexaenoic acid served as substrates for the enzyme. In the case of 8,11,14-eicosatrienoic acid (3) and 5,8,11,14,17-eicosapentaenoic acid (4) it was later confirmed that oxygenation occurred at C-12 (ω9 position) as expected.

The present paper, which is concerned with the metabolism of 6,9,12-octadecatrienoic acid in human platelets, demonstrates the presence in platelets of a lipoxygenase activity that efficiently catalyzes oxygenation at the ω6 position, in addition to the cyclooxygenase and ω9-lipoxygenase activities previously recognized.

MATERIALS AND METHODS

[1-<sup>14</sup>C]6,9,12-Octadecatrienoic acid was prepared by incubation of [1-<sup>14</sup>C]stearic acid with growing cultures of *Tetrahymena pyriformis*, strain GL (generously provided by Dr. O. Vestergaard, Aarhus Universitet, Denmark) at room temperature for 44-

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48 h (cf. 5). The chemical and radiochemical purity of the acid obtained was 95% or better and the specific radioactivity was 1 Ci/mol. Unlabeled 6,9,12-octadecatrienoic acid was purchased from Nu-Chek-Prep., Inc., Elysian, Minn.

Suspensions of human platelets ( $1-1.5 \times 10^6/\mu\text{l}$ ) in 0.154 M NaCl, 25 mM tris(hydroxymethyl)aminomethane, and 0.2 mM disodium EDTA; pH 7.4 were prepared as previously described (4).

Incubations of [ $1-^{14}\text{C}$ ]6,9,12-octadecatrienoic acid ( $7-1103 \mu\text{M}$ ) with platelet suspensions (10-15 ml) were carried out at  $37^\circ$  for 20 min with magnetic stirring. When inhibitors were included, these were incubated with the platelet suspension for 10 min at  $37^\circ$  (0.5 mM aspirin) or 2 min at  $37^\circ$  (10  $\mu\text{M}$  indomethacin; 34  $\mu\text{M}$  5,8,11,14-eicosatetraynoic acid) prior to addition of substrate. The incubations were stopped by the addition of 5 vol. of ethanol and extracted with diethyl ether. The residue obtained after evaporation of the solvent was treated with diazomethane and subjected to thin layer radiochromatography (solvent, diethyl ether - hexane (1:1; v/v)). Material recovered following thin layer chromatography was further purified by straight-phase high performance liquid chromatography using a column of Nucleosil 5  $\mu$  (250x4.6 mm) and isopropanol - hexane (1:99; v/v) as the solvent.

Gas-liquid chromatography was carried out with an F&M Biomedical gaschromatograph, model 402. The stationary phases used were 10% SP-2330 on Supelcoport (analysis of poly-unsaturated fatty acid esters), 1% SE-30 on GasChrom Q (analysis of derivatives of hydroxy acids), and 5% QF-1 on GasChrom Q (analysis of ozonolysis products). Mass spectrometry in combination with gas-liquid chromatography was performed with an LKB 9000 instrument equipped with a column of 1% OV-1 on Chromosorb W.

Catalytic hydrogenation was carried out by stirring the unsaturated compound (methyl ester); 10-50  $\mu\text{g}$ ) with 2 mg of platinum oxide in 2 ml of methanol for 15 min under an atmosphere of  $\text{H}_2$ .

## RESULTS

Identification of Products. Four major peaks of radioactivity were observed upon thin-layer radiochromatographic analysis of the esterified product isolated following incubation of [ $1-^{14}\text{C}$ ] 6,9,12-octadecatrienoic acid with human platelets (Fig. 1). The peak close to the solvent front ( $R_F = 0.94$ ) corresponded to substrate remaining unconverted. Material present in the peak at the origin was rechromatographed in a more polar solvent system (methanol - diethyl ether (3:97; v/v)) using the methyl esters of thromboxane  $\text{B}_2$  and prostaglandins  $\text{E}_2$  and  $\text{F}_{2\alpha}$  as references. No peak of radioactivity appeared in the region of the reference compounds. Instead, the labeled material remained at the origin suggesting its identity with phospholipid(s) that had incorporated the labeled 6,9,12-octadecatrienoic acid.

Material present in peak A (Fig. 1) showed strong ultraviolet absorption with  $\lambda_{\text{max}}^{\text{EtOH}} = 233 \text{ nm}$  indicating the presence of a pair

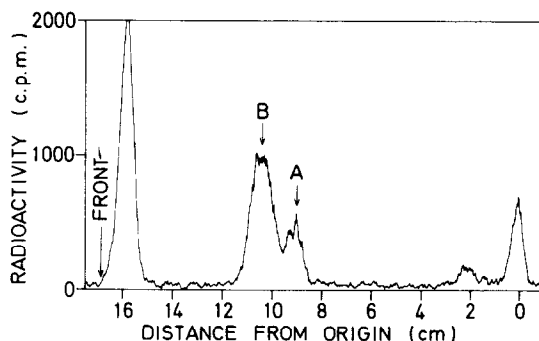


Fig. 1. Thin layer radiochromatogram of esterified product obtained after incubation of 133  $\mu\text{g}$  of  $[1\text{-}^{14}\text{C}]6,9,12\text{-octadecatrienoic}$  acid with 12 ml of suspension of human platelets ( $1.22 \times 10^6$  platelets/ $\mu\text{l}$ ) at  $37^\circ$  for 20 min. Solvent system: diethyl ether - hexane (1:1; v/v).

of conjugated trans double bonds. Analysis of the  $\text{Me}_3\text{Si}^1$  derivative by GC-MS showed a major peak with an equivalent chain length corresponding to C-17.5. The mass spectrum showed prominent ions at  $m/e$  340 (M), 325 (M-15; loss of  $\cdot\text{CH}_3$ ), 309 (M-31; loss of  $\cdot\text{OCH}_3$ ), 269 (M-71; loss of  $\cdot(\text{CH}_2)_4\text{CH}_3$ ), 250 (M-90; loss of  $\text{Me}_3\text{SiOH}$ ), 225 (M-115; loss of  $\cdot(\text{CH}_2)_4\text{COOCH}_3$ ), and 147 (269-(90+32)). This suggested that Compound A was the methyl ester of 10-hydroxy-6,8-pentadecadienoic acid. Mass-spectrometric analysis of the  $\text{Me}_3\text{Si}$  derivative of the hydrogenated methyl ester (C-17.0) showed ions inter alia at  $m/e$  329 (M-15; loss of  $\cdot\text{CH}_3$ ), 273 (M-71; loss of  $\cdot(\text{CH}_2)_4\text{CH}_3$ ), 244 (M-100; rearrangement followed by loss of  $\text{O=HC}-(\text{CH}_2)_4\text{CH}_3$ ), and 173 ( $\text{Me}_3\text{SiO}^+=\text{CH}-(\text{CH}_2)_4\text{CH}_3$ ) demonstrating a methyl pentadecanoate carrying a hydroxyl group at C-10. Oxidative ozonolysis of the (-)menthoxycarbonyl derivative (cf. 6) of Compound A afforded methyl hydrogen adipate and the (-)menthoxycarbonyl derivative of 2L-hydroxyheptanoic acid. On basis of these results Compound A was assigned the structure 10L<sub>S</sub>-hydroxy-6,8-pentadecadienoic acid (methyl ester). The abbreviation 10-HPD is proposed for this hydroxy acid.

Material present in peak B (Fig. 1) was separated into two compounds, i.e. Compounds B<sub>I</sub> and B<sub>II</sub>, by straight-phase high performance liquid chromatography (Fig. 2). Both compounds showed

<sup>1</sup>The abbreviations used are:  $\text{Me}_3\text{Si}$ , trimethylsilyl; GC-MS, gas-liquid chromatography - mass spectrometry; 10-HPD, 10L<sub>S</sub>-hydroxy-6,8-pentadecadienoic acid; 10-HOT, 10L<sub>S</sub>-hydroxy-6,8,12-octadecatrienoic acid; 13-HOT, 13L<sub>S</sub>-hydroxy-6,9,11-octadecatrienoic acid.

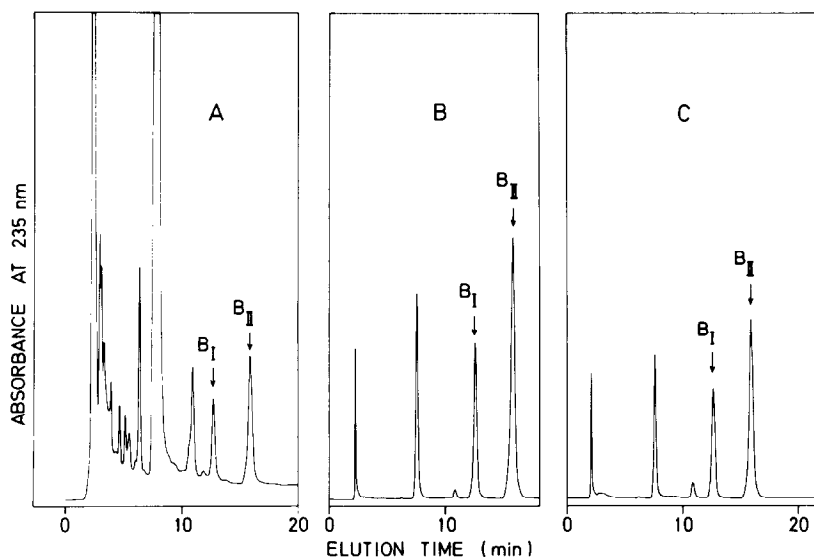


Fig. 2. Straight-phase high performance liquid chromatography of material present in peak B, Fig. 1. Column, Nucleosil 5  $\mu$  (250x4.6 mm); solvent, isopropanol - hexane (1:99; v/v); flow rate, 1.5 ml/min. Concentrations of incubated 6,9,12-octadecatrienoic acid: in "A", 2  $\mu$ g/ml; in "B", 45  $\mu$ g/ml; in "C", 18  $\mu$ g/ml.

strong ultraviolet absorption with  $\lambda_{\text{max}}^{\text{EtOH}} = 235 \text{ nm}$  indicating the presence of a pair of cis/trans conjugated double bonds. The mass spectrum of the  $\text{Me}_3\text{Si}$  derivative of Compound  $\text{B}_I$  (C-19.6) showed ions of high intensity at  $m/e$  380 (M), 309 (M-71; loss of  $\cdot(\text{CH}_2)_4\text{CH}_3$ ), 290 (M-90; loss of  $\text{Me}_3\text{SiOH}$ ), 225 ( $[(\text{CH}=\text{CH})_2\text{-CH}(\text{OSiMe}_3)\text{-(CH}_2)_4\text{CH}_3]^+$ ), and 173 ( $\text{Me}_3\text{SiO}^+=\text{CH-(CH}_2)_4\text{-CH}_3$ ) suggesting a methyl 13-hydroxyoctadecatrienoate. The mass spectrum of the  $\text{Me}_3\text{Si}$  derivative of hydrogenated Compound  $\text{B}_I$  (C-20.0) showed prominent ions at  $m/e$  371 (M-15; loss of  $\cdot\text{CH}_3$ ), 355 (M-31; loss of  $\cdot\text{OCH}_3$ ), 339 (M-(15+32)), 315 (M-71; loss of  $\cdot(\text{CH}_2)_4\text{CH}_3$ ), 286 (M-100; rearrangement followed by loss of  $\text{O=HC-(CH}_2)_4\text{CH}_3$ ), and 173 ( $\text{Me}_3\text{SiO}^+=\text{CH-(CH}_2)_4\text{CH}_3$ ) in agreement with a methyl octadecanoate carrying a hydroxyl group at C-13. Oxidative ozonolysis performed on the (-)-menthoxy carbonyl derivative of Compound  $\text{B}_I$  afforded methyl hydrogen adipate and the (-)-menthoxy carbonyl derivative of 2L-hydroxyheptanoic acid (Fig. 3). This result taken together with the ultraviolet data demonstrated that the double bonds of Compound  $\text{B}_I$  were located at  $\Delta^6$ ,  $\Delta^9$ , and  $\Delta^{11}$ , and that the hydroxyl group at C-13 had the  $\text{L}_S$  configuration. Compound  $\text{B}_I$  was thus methyl 13 $\text{L}_S$ -hydroxy-6,9,11-octadecatrienoate. The abbreviation 13-HOT is proposed for the parent acid. Compound  $\text{B}_{II}$  was also converted into the  $\text{Me}_3\text{Si}$  derivative and

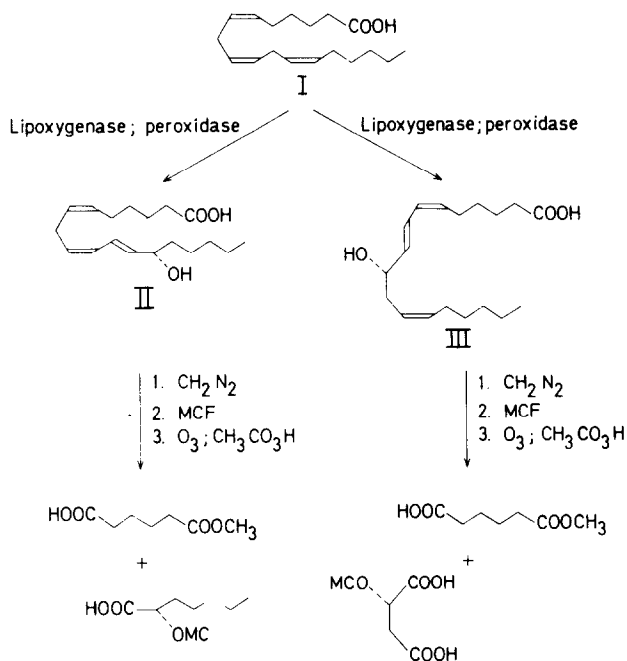


Fig. 3. Chemical degradation of 10- and 13-hydroxyoctadecatrienoates.

"I", 6,9,12-Octadecatrienoic acid; "II", 13 $\text{L}_\text{S}$ -hydroxy-6,9,11-octadecatrienoic acid; "III", 10 $\text{L}_\text{S}$ -hydroxy-6,8,12-octadecatrienoic acid; "MCF", (-)-menthylchloroformate; "MC", (-)-menthoxy-carbonyl.

analyzed by GC-MS (C-19.6). The mass spectrum showed a prominent ion at  $m/e$  269 ( $\text{M}-111$ ; loss of  $\cdot\text{CH}_2-\text{CH}=\text{CH}-(\text{CH}_2)_4\text{CH}_3$ ) as well as ions of lower intensity at  $m/e$  365 ( $\text{M}-15$ ; loss of  $\cdot\text{CH}_3$ ), 349 ( $\text{M}-31$ ; loss of  $\cdot\text{OCH}_3$ ), 290 ( $\text{M}-90$ ; loss of  $\text{Me}_3\text{SiOH}$ ), 179 ( $269-90$ ), and 147 ( $179-32$ ). Catalytic hydrogenation of Compound B $_{\text{II}}$  afforded methyl 10-hydroxyoctadecanoate as judged by mass-spectrometric analysis of the  $\text{Me}_3\text{Si}$  derivative (prominent ions at, inter alia,  $m/e$  273 ( $\text{M}-113$ ; loss of  $\cdot(\text{CH}_2)_7\text{CH}_3$ ), 244 ( $\text{M}-142$ ; rearrangement followed by loss of  $\text{O}=\text{HC}-(\text{CH}_2)_7\text{CH}_3$ ), and 215 ( $\text{Me}_3\text{SiO}^+=\text{CH}-(\text{CH}_2)_7\text{CH}_3$ )). Oxidative ozonolysis of the (-)-menthoxycarbonyl derivative gave rise to methyl hydrogen adipate and the (-)-menthoxycarbonyl derivative of L-malic acid (Fig. 3). The data thus showed that Compound B $_{\text{II}}$  was identical with methyl 10 $\text{L}_\text{S}$ -hydroxy-6,8,12-octadecatrienoate. The proposed abbreviation for the parent acid is 10-HOT.

Effect of Inhibitors. Conversion of 6,9,12-octadecatrienoic acid into 10-HPD was completely blocked in the presence of either 0.5 mM aspirin or 10  $\mu\text{M}$  indomethacin. Formation of the two  $\text{C}_{18}$  hydroxy acids was not affected by these agents. This shows that

10-HPD was formed by the cyclooxygenase pathway (by elimination of malonaldehyde from the  $C_{18}$ -prostaglandin endoperoxide,  $9\alpha$ ,  $11\alpha$ -epidioxy- $15L_G$ -hydroxy-2,3-dinorprost-13-enoic acid) whereas the two latter compounds were formed by action of lipoxygenase(s). In agreement with this, the cyclooxygenase and lipoxygenase inhibitor, 5,8,11,14-eicosatetraynoic acid ( $34 \mu M$ ), blocked formation of all three compounds.

Effect of Precursor Concentration. [ $1-^{14}C$ ]6,9,12-Octadecatrienoic acid at concentrations 7-1103  $\mu M$  was incubated with suspensions of human platelets. The methyl esters of 10-HOT and 13-HOT were isolated by thin layer chromatography (peak B, Fig. 1) and the ratio 10-HOT/13-HOT was determined by straight-phase high performance liquid chromatography (Fig. 2). As seen from Table I the ratio between the two isomers was essentially constant, i.e. 10-HOT/13-HOT = 1.94-2.11.

#### DISCUSSION

The present study shows that 6,9,12-octadecatrienoic acid is converted into cyclooxygenase as well as lipoxygenase products when incubated with human platelets. This is in agreement with previous results obtained with arachidonic acid (1,2), 8,11,14-eicosatrienoic acid (3) and 5,8,11,14,17-eicosapentaenoic acid (4). However, in the case of 6,9,12-octadecatrienoic acid the only cyclooxygenase product that could be isolated was a  $C_{15}$  monohydroxy acid,  $10L_G$ -hydroxy-6,8-pentadecadienoic acid. Thromboxane- or prostaglandin-like material could not be detected. Another difference concerns the lipoxygenase pathway: the  $C_{20}$  polyenoic acids were each converted into a single monohydroxy acid by oxygenation at C-12 ( $\omega 9$ ) (1-4) whereas 6,9,12-octadecatrienoic acid gave rise to two isomeric hydroxy acids by oxygenation at  $\omega 9$  and  $\omega 6$ . The ratio between the two isomers was about 2.0. The finding that this ratio was not significantly changed when substrate concentrations ranging from 7 to 1103  $\mu M$  were used suggests action of a single platelet lipoxygenase with dual positional specificity. However, further studies are needed to determine whether one or two lipoxygenases are involved. After the present paper was completed a study appeared on hydroxy acid synthesis from 4,7,10,13,16,19-docosahexaenoic acid by human platelets (7). Two major isomers, i.e.  $14(\omega 9)$ - and  $11(\omega 12)$ -hydroxydocosahexaenoic acids were isolated. In that study the ratio between the two isomers was found to vary with respect to

substrate concentration (14-/11-hydroxy isomer at 6.8  $\mu$ M docosa-hexaenoic acid, 4.2; 14-/11-hydroxy isomer at 54  $\mu$ M docosa-hexaenoic acid, 1.9; ref. 7).

Although the present paper for the first time shows efficient (Table I) conversion of a poly-unsaturated fatty acid into an  $\omega$ 6-monohydroxy acid by action of  $\omega$ 6-lipoxygenase activity in platelets two previous studies in which small amounts of dihydroxy acids inter alia oxygenated in the  $\omega$ 6 position should be mentioned. These are 1) the study on the transformations of 8,11,14-eicosatrienoic acid in platelets (3) in which small amounts of 14,15( $\omega$ 6)-dihydroxy-8,10,12-eicosatrienoic acid and 8,15( $\omega$ 6)-dihydroxy-9,11,13-eicosatrienoic acid were tentatively identified; and 2) a recent report that showed release of small amounts of 14,15( $\omega$ 6)-dihydroxy-5,8,10,12-eicosatetraenoic acid from platelets upon exposure to bradykinin (8).

The significance of the  $\omega$ 6-lipoxygenase pathway in platelets is unknown, however, recent studies on the biological activity of eg. 14,15-dihydroxy-5,8,10,12-eicosatetraenoic acid (9) make further investigation of the platelet  $\omega$ 6-lipoxygenase pathway and its products of interest.

Table I

Effect of precursor concentration on the yield and proportion of hydroxy-octadecatrienoate isomers. Each incubation was carried out with 10 ml of platelet suspension preincubated with 0.5 mM aspirin. "18:3" = 6,9,12-octadecatrienoic acid.

18:3 incubated ( $\mu$ g)	10-HOT+13-HOT isolated * ( $\mu$ g)	10-HOT/13-HOT**
<u>Donor 1</u>		
20	N.D.	1.97
50	24	1.94
180	82	2.04
450	220	2.11
<u>Donor 2</u>		
67	33	2.00
567	264	2.09
3067	313	2.07

\* The amounts were determined by ultraviolet spectrometry after isolation by thin layer chromatography (see Fig. 1).

\*\* The ratios were determined by straight-phase high performance liquid chromatography (see Fig. 2).

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## REFERENCES

1. Hamberg, M., and Samuelsson, B. (1974) Proc. Nat. Acad. Sci. U.S.A. 71, 3400-3404.
2. Nugteren, D.H. (1975) Biochim. Biophys. Acta 380, 299-307.
3. Falardeau, P., Hamberg, M., and Samuelsson, B. (1976) Biochim. Biophys. Acta 441, 193-200.
4. Hamberg, M. (1980) Biochim. Biophys. Acta 618, 389-398.
5. Hamberg, M., and Samuelsson, B. (1967) J. Biol. Chem. 242, 5336-5343.
6. Hamberg, M. (1971) Anal. Biochem. 43, 515-526.
7. Avelldano, M.I., and Sprecher, H. (1983) J. Biol. Chem. 258, 9339-9343.
8. Wong, P.Y-K., Westlund, P., Granström, E., Hamberg, M., Chao, P.H-W., and Samuelsson, B. (1983) Thrombos. Haemostas. 50(1), Abstract No. 389.
9. Lundberg, U., Serhan, C.N., Weissmann, G., and Samuelsson, B. (1983) to be published.